**rac-(Z)-2-(2-Thienylmethylene)-1-aza-bicyclo[2.2.2]octan-3-ol**

Vijayakumar N. Sonar, M. Venkatraj, Sean Parkin and Peter A. Crooks

*Department of Pharmaceutical Sciences, College of Pharmacy, University of Kentucky, Lexington, KY 40536, USA, and* Department of Chemistry, University of Kentucky, Lexington, KY 40506, USA

Correspondence e-mail: pcrooks@email.uky.edu

Received 30 May 2007
Accepted 10 July 2007
Online 26 July 2007

The asymmetric unit of the racemic form of the title compound, C_{12}H_{12}NOS, contains four crystallographically independent molecules. The olefinic bond connecting the 2-thienyl and 1-aza-bicyclo[2.2.2]octan-3-ol moieties has Z geometry. Strong hydrogen bonding occurs in a directed cooperative O—H⋯O—H⋯O—H⋯O pattern that influences the conformation of the molecules. Co-operative C—H⋯π interactions between thienyl rings are also present. The average dihedral angle between adjacent thienyl rings is 87.09 (4)°.

**Comment**

Radiation therapy prevents cancer cells from growing and dividing, but can also damage normal cells. Radio-sensitizers are drugs that sensitize cancer cells to radiation therapy. Recent research in our laboratory has focused on the discovery of molecules that can either sensitize a tumour to radiation without affecting healthy tissue, or that can shield normal cells from radiation damage. Recently, we have reported (Sekhar et al., 2007) the radio-sensitizing activity of N-aryl-substituted rac-(Z)-2-(1H-indol-3-ylmethylene)-1-aza-bicyclo[2.2.2]octan-3-ols. In addition to these indole analogues, we have also synthesized 1-aza-bicyclo[2.2.2]octan-3-ols linked to non-indolic systems, in order to compare their radio-sensitizing activities with those of the indole analogues. The title compound, (I), was prepared by the base-catalyzed reaction of thiophene-2-carbaldehyde with 1-aza-bicyclo[2.2.2]octan-3-one and subsequent reduction to the corresponding secondary alcohol. In order to confirm the double-bond geometry, and to determine the molecular conformation in the crystal structure, the X-ray analysis of compound (I) has been carried out, and the results are presented here.

The asymmetric unit of (I) contains four crystallographically independent molecules, (IA)–(ID), with almost identical bond lengths and angles. The molecular structure and atom-numbering scheme of molecule (IA) are shown in Fig. 1, and selected bond lengths and angles are listed in Table 1.

![Figure 1](image-url)

The title molecule comprises a 1-aza-bicyclo[2.2.2]octan-3-ol moiety and a 2-thienyl group linked by a C5=C6 bond that has Z geometry. In this Z isomer, the C1—C5 bond is trans with respect to the C6—C10 bond. The double bond has a nearly planar atomic arrangement, since the r.m.s. deviation from the best plane passing through atoms N1, C6, C10, C5 and C1 is 0.0061 (10) Å for (IA), 0.0972 (10) Å for (IB), 0.1943 (10) Å for (IC) and 0.2500 (10) Å for (ID). The geometries of the four molecules in the asymmetric unit are essentially the same, so only molecule A is discussed in detail here.

The bond angles around atoms C5A and C6A deviate from 120° and this is in general agreement with our earlier observation for a similar type of compound (Sonar et al., 2004). The angles N1A—C6A—C10A, C6A—C5A—C1A and C5A—C6A—C10A [114.02 (15), 126.55 (17) and 125.26 (17)°, respectively] are distorted as a consequence of the strain induced by the double-bond linkage at C5A=C6A. The aza-bicyclic system exhibits small distortions in geometry compared with standard tetrahedral values. The angles C6A—N1A—C7A, C10A—C9A—C11A and C9A—C11A—C12A [107.28 (14), 107.41 (15) and 107.79 (14)°, respectively] are all smaller than the ideal tetrahedral value of 109.5°, while the angles N1A—C7A—C8A, O1A—C10A—C9A and N1A—C12A—C11A [111.32 (15), 113.84 (14) and 112.05 (14)°, respectively] are all slightly larger than the tetrahedral value. The C4A—C1—C5—C6 torsion angles [170.67 (19),
organic compounds

Figure 2
A view of molecules (I)–(IV). H atoms have been omitted for clarity, except for those involved in \( \pi \) interactions and O–H···O hydrogen bonds (dashed lines).

173.55 (19), 173.41 (19) and 170.85 (19)° for molecules A, B, C and D, respectively) indicate a deviation of the 2-thienyl ring from the plane of the double bond connecting it to the azabicyclic ring in these molecules. This small deviation facilitates a weak conjugation between the C5=C6 double bond and the 2-thienyl ring. This is also evident from the C1–C5 bond lengths [1.456 (3), 1.455 (3), 1.458 (3) and 1.451 (3) Å in molecules A, B, C and D, respectively].

Strong hydrogen bonds and weaker C–H···\( \pi \) interactions (Fig. 2) present in the crystal structure of (I) effectively group sets of four molecules together into roughly cylindrical units. Within each group, all molecules possess the same absolute configuration and are related to other groups of opposite hand via crystallographic inversion centres. The donor–acceptor direction of the \( R_1^2(8) \) hydrogen-bonded motif comprising O1A–H1A···O1B–H1B···O1C–H1C···O1D–H1D···O1A (Table 2) is opposite from the directionality around the C–H···\( \pi \)-linked thiienyl rings [(ring A)–H4A···(ring D)–H4D···(ring C)–H4C···(ring B)–H4B···(ring A)]. These C–H···\( \pi \) interactions, C4A···(ring B), C4B···(ring C), C4C···(ring D) and C4D···(ring A), with distances of 3.551 (2), 3.531 (2), 3.554 (2) and 3.565 (2) Å, respectively, effectively lock the thiophene rings into a single conformation. The dihedral angles between successive rings are \( A/B = 89.24 \) (4)°, \( B/C = 86.64 \) (4)°, \( C/D = 85.08 \) (4)° and \( D/A = 87.39 \) (4)°.

Experimental
The title compound was prepared according to the previously reported procedure of Sonar et al. (2004). Crystalisation from ethyl acetate afforded colourless crystals. \(^1\)H NMR (CDCl₃): \( \delta \) 1.47–1.89 (m, 5H), 2.05 (s, 1H), 2.74–3.00 (m, 4H), 4.34 (s, 1H), 6.59 (s, 1H), 6.97 (d, 2H), 7.25 (d, 1H); \(^{13}\)C NMR (CDCl₃): \( \delta \) 19.07, 25.37, 31.09, 46.64, 47.77, 70.61, 116.83, 125.47, 126.95, 127.49, 138.53, 149.20.

Crystal data

\[ \text{C}_2\text{H}_7\text{NO}_3 \]

\( M_r = 221.32 \)

Monoclinic, \( P_2_1/n \)

\( a = 14.7933 \) (3) Å

\( b = 20.0349 \) (4) Å

\( c = 14.8243 \) (3) Å

\( \beta = 90.2820 \) (7)°

\( V = 4393.61 \) (15) Å³

\( Z = 16 \)

Mo Kα radiation

\( \mu = 0.27 \) mm⁻¹

\( T = 90.0 \) (2) K

\( \rho = 0.38 \times 0.35 \times 0.03 \) mm

Data collection

Nonius KappaCCD area-detector diffractometer

37339 measured reflections

10043 independent reflections

\( 7113 \) reflections with \( I > 2\sigma(I) \)

\( R_{int} = 0.058 \)

Refinement

\( R[F^2 > 2\sigma(F^2)] = 0.044 \)

\( wR(F^2) = 0.114 \)

\( S = 1.04 \)

545 parameters

H-atom parameters constrained

\( \Delta\rho_{	ext{max}} = 0.31 \) e Å⁻³

\( \Delta\rho_{	ext{min}} = -0.34 \) e Å⁻³

Table 1

Selected geometric parameters (Å, °).

<table>
<thead>
<tr>
<th>Bond</th>
<th>Length (Å)</th>
<th>Angle (°)</th>
</tr>
</thead>
<tbody>
<tr>
<td>O1A–C10A</td>
<td>1.431 (2)</td>
<td>O1C–C10C</td>
</tr>
<tr>
<td>C6A–C10A</td>
<td>1.523 (2)</td>
<td>C6C–C10C</td>
</tr>
<tr>
<td>S1B–C1B</td>
<td>1.7348 (19)</td>
<td>S1D–C1D</td>
</tr>
<tr>
<td>O1B–C10B</td>
<td>1.435 (2)</td>
<td>O1D–C1D0</td>
</tr>
<tr>
<td>C6B–C10B</td>
<td>1.516 (2)</td>
<td>C6D–C1D0</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Bond</th>
<th>Length (Å)</th>
<th>Angle (°)</th>
</tr>
</thead>
<tbody>
<tr>
<td>C6A–C5A–C1A</td>
<td>126.55 (17)</td>
<td>C6C–C5C–C1C</td>
</tr>
<tr>
<td>C5A–C6A–C10A</td>
<td>125.26 (17)</td>
<td>C5C–C6C–C10C</td>
</tr>
<tr>
<td>N1A–C6A–C10A</td>
<td>114.02 (15)</td>
<td>N1C–C6C–C10C</td>
</tr>
<tr>
<td>C5B–C5B–C1B</td>
<td>127.41 (17)</td>
<td>C5C–C5C–C1C</td>
</tr>
<tr>
<td>C5B–C5B–C1B</td>
<td>125.22 (17)</td>
<td>C5D–C5D–C1D</td>
</tr>
<tr>
<td>N1B–C5B–C10B</td>
<td>114.17 (15)</td>
<td>N1D–C5D–C1D</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Bond</th>
<th>Length (Å)</th>
<th>Angle (°)</th>
</tr>
</thead>
<tbody>
<tr>
<td>S1A–C1A</td>
<td>1.7350 (19)</td>
<td>S1C–C1C</td>
</tr>
<tr>
<td>S1B–C1B</td>
<td>1.7348 (19)</td>
<td>S1D–C1D</td>
</tr>
<tr>
<td>S1B–C1B</td>
<td>1.7348 (19)</td>
<td>S1D–C1D</td>
</tr>
</tbody>
</table>

Table 2

Hydrogen-bond geometry (Å, °).

\( \text{D} \cdot \cdot \cdot \text{H} \cdot \cdot \cdot \text{A} \quad \text{D} \cdot \cdot \cdot \text{H} \cdot \cdot \cdot \text{A} \quad \text{D} \cdot \cdot \cdot \text{H} \cdot \cdot \cdot \text{A} \quad \text{D} \cdot \cdot \cdot \text{H} \cdot \cdot \cdot \text{A} \)

<table>
<thead>
<tr>
<th>Bond</th>
<th>Length (Å)</th>
<th>Angle (°)</th>
</tr>
</thead>
<tbody>
<tr>
<td>O1A–H1A···O1B</td>
<td>0.84</td>
<td>1.97</td>
</tr>
<tr>
<td>O1B–H1B···O1C</td>
<td>2.02</td>
<td>2.02</td>
</tr>
<tr>
<td>O1C–H1C···O1D</td>
<td>1.97</td>
<td>1.97</td>
</tr>
<tr>
<td>O1D–H1D···O1A</td>
<td>1.97</td>
<td>1.97</td>
</tr>
</tbody>
</table>

H atoms were found in a difference Fourier map and subsequently placed in idealized positions with constrained C–H distances of 0.95 (Csp³–H), 0.99 (CH₃) and 1.00 Å (CH), and with O–H = 0.84 Å. \( U_{eq}(H) \) values were set at either 1.2\( U_{eq} \) or 1.5\( U_{eq} \) (OH only) of the parent atom.

Data collection: COLLECT (Nonius, 1999); cell refinement: SCALEPACK (Otwinski & Minor, 1997); data reduction: DENZO-SMN (Otwinski & Minor, 1997); program(s) used to solve structure: SHELXL97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: XP in SHELXTL/PC (Sheldrick, 1995); software used to prepare material for publication: SHELXL97 and local procedures.
This investigation was supported by the National Cancer Institute (grant No. PPG NIH/NCI PO1 CA104457-01 A1).

Supplementary data for this paper are available from the IUCr electronic archives (Reference: HH3044). Services for accessing these data are described at the back of the journal.

References


